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A Stereoselective Synthesis of the C-10 to C-18 (Right-Half) Fragment of Mycalamides Employing Lewis Acid Promoted Intermolecular Aldol Reaction

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ABSTRACT



Beginning with p-mannitol, a stereoselective synthesis of the right-half segment of the mycalamides has been accomplished by employing Lewis acid catalyzed intermolecular aldol reaction and oxypalladation as the key steps.

Mycalamides A $(1)^1$ and B (2),² isolated from a New Zealand sponge of the genus *Mycale*, are potent antiviral and antitumor compounds (Figure 1).³ On top of that, mycalamides A (1) and B (2) show strong immunosuppressive activity.⁴ The mycalamides are structurally related to onnamides⁵ and theopederins⁶ and have a highly functionalized system in which an *exo*-olefinic tetrahydropyran connects with a trioxa-*cis*-decalin ring through a hydroxyacetamide moiety. The unique structural features of the mycalamides coupled with their biological properties have provided

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Mycalamide A (1): R=H Mycalamide B (2): R=Me

Figure 1.

motivation for the development of synthetic strategies toward these marine natural products.⁷ Herein we would like to present a stereoselective synthesis of the right segment (C-10 to C-18) of the mycalamides.

⁽⁷⁾ **Total syntheses of the mycalamides:** (a) Hong, C. Y.; Kishi, Y. *J. Org. Chem.* **1990**, *55*, 4242. (b) Nakata, T.; Matsukura, H.; Jian, D.; Nagashima, H. *Tetrahedron Lett.* **1994**, *35*, 8229. (c) Kocienski, P. J.; Narquizian, R.; Raubo, P.; Smith, C.; Boyle, F. T. *Synlett* **1998**, 869.

Our retrosynthetic analysis for the right segment, **3**, of the mycalamides is shown in Scheme 1. The first key reaction



of the synthesis is the oxypalladation of **5** for the stereoselective construction of the trioxa-*cis*-decalin ring system **4**, which already has the right-half skeleton **3** of the mycalamides and suitable functionalities. A series of functional group modifications in **6** would lead to **5**. The second key step of the present synthesis is the Lewis acid promoted intermolecular aldol reaction of aldehyde **7** with dimethylketene methyl trimethylsilyl acetal.

The requisite aldehyde **12** for a Lewis acid promoted intermolecular aldol reaction was prepared in a stereoselective manner as depicted in Scheme 2. Thus, dihydroxylation of olefin **8**,⁸ synthesized from D-mannitol, with a catalytic amount of osmium tetroxide in the presence of *N*-methylmorpholine oxide (1.5 equiv) afforded the corresponding diol (α : $\beta = 8:1$),⁹ which was subjected to monosilylation to produce the desired α -alcohol **9** as a single stereoisomer after recrystallization. Hydrolysis of the acetal group of **9** followed by monoesterification with pivaloyl chloride furnished compound **10**, which was transformed into alcohol **11** by 1,3-dioxane formation¹⁰ and DIBALH reduction of the pivaloate ester. Alcohol **11** was oxidized to aldehyde **12** without epimerization under Swern conditions.¹¹

We initially investigated the intermolecular aldol reaction¹² of aldehyde **12** with the dimethylketene acetal in the presence



^{*a*} Reagents and conditions: (a) OsO₄ (cat.), NMO (100%, α: β = 8:1); (b) *t*-Bu(Ph)₂SiCl, DMAP, Et₃N (51% after recrystallization); (c) AcOH-THF-H₂O (3:1:1), 55 °C (98%); (d) *t*-BuCOCl, pyridine (89%); (e) (MeO)₂CH₂, P₂O₅ (88%); (f) DIBALH, 0 °C (99%); (g) (COCl)₂, DMSO, Et₃N, -78 to -40 °C.

of 10 mol % of Yb(OTf)₃ (Table 1, entry 1). Under the above conditions, TMS ether **13** was obtained in 62% yield, as a major product, together with alcohol **14** (15% yield). When

Table 1. Lewis Acid Promoted Intermolecular Aldol Reaction^a

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R ¹ 0	OBn CH	O Me Lewis a CH ₂ C	OTMS OMe acid I ₂ R ¹ O	OBr	OR ² C Me Me	O₂Me
entry	Lewis acid	equiv- alent	temp, °C	$13, R^2 = TMS^b$	14 , R^2 = H^b	12 ^c
1	Yb(OTf) ₃	10 mol %	rt	62%	15%	
2	Sc(OTf) ₃	10 mol %	-78 to rt		19%	19%
3	Cu(OTf) ₃	10 mol %	rt		41%	25%
4	InCl ₃	10 mol %	rt	41%	21%	
5	TiCl ₄	150 mol %	-78		63%	

 $^a\,\rm R^1=Si(Ph)_{2}t\text{-}Bu.$ b Yield for 2 steps from the alcohol 11. c Recovered starting material.

Sc(OTf)₃ and Cu(OTf)₂ were used as Lewis acid, a considerable amount of the starting material **12** was recovered in addition to **14** (entries 2 and 3). Subjecting **12** to InCl₃ at room temperature led to the coupled products (**13** and **14**, with a combined yield greater than 60% yield, entry 4). Unlike the above results, only **14** was isolated when employing TiCl₄ as Lewis acid¹³ (entry 5).

The stereochemistry of hydroxyester 14 was determined after the conversion of 14 to the bicyclic compound 19 as

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⁽¹³⁾ We evaluated Lewis acids such as TMSOTf, Eu(OTf)₃, Gd(OTf)₃, and Y(OTf)₃; however, all the Lewis acids tested gave low yields.

shown in Figure 3. The stereochemical outcome of the present aldol reaction might be rationalized by nucleophilic attack from less hindered face (Figure 2). With stereoselective



Figure 2.

induction of the side chain established, we next focused on the formation of the second ring of 4 (Scheme 1). The required substrate 16 for oxypalladation¹⁴ was readily prepared via a six-step sequence from 14 (Scheme 3).



^{*a*} Reagents and conditions: (a) NaH, MeI, 0 °C to rt (95%); (b) DIBALH, 0 °C (100%); (c) SO₃·Py DMSO, Et₃N (97%); (d) Ph₃PEtBr, BuLi, 0 °C (100%); (e) liquid NH₃, Li, -78 °C; (f) DDQ (73% for two steps).

Methylation of alcohol **14** followed by DIBALH reduction and oxidation provided aldehyde **15**, which was allowed to undergo a Wittig reaction to give the corresponding olefin. After reductive deprotection of the benzyl group, the phenyl moiety on the silicon, partially reduced, was reoxidized with DDQ to afford **16**.

Upon treatment of olefin **16** with $PdCl_2$ in the presence of CuCl in a mixture of DMF and water under a oxygen atmosphere, the oxypalladation proceeded smoothly, producing cyclized products (**17** and **18**), in 89% yield, as a 3:1 separable mixture (Table 2). The result from the next example was intriguing. Switching catalyst to $Pd(OAc)_2^{15}$

	17		18	19						
entry	Pd	additive	solvent	17	18	19				
1 2	PdCl ₂ Pd(Oac) ₂	CuCl, O ₂	DMF-H ₂ O DMSO	67%	22% 84%	trace 9%				
^{<i>a</i>} $\mathbf{R} = \mathrm{CH}_{2}\mathrm{OSi}(\mathrm{Ph})_{2}t$ -Bu.										

led to **18** (84%) together with **19** (9%). The structures of **18** and **19** were confirmed by their NOE experiments as shown in Figure 3. Since the formation of the second ring proved



to be more difficult than expected,¹⁶ we decided to apply the experimental result (entry 1) of the oxypalladation of compound **16** to the construction of enol ether **17**. After transformation of **15** into the requisite substrate in three steps (74% overall yield), the oxypalladation was carried out under the reaction conditions similar to those of entry 1 (Table 2) to provide the desired enol ether **20** in 83% yield.

With the skeletal framework assembled, completion of the synthesis of the right segment of the mycalamides would require the installation of the side chain and the carbamate group. Compound **20** was subjected to ozonolysis followed by DIBALH reduction and acetylation, generating the corresponding lactol, which was exposed to allyltrimethyl-silane in the presence of BF₃•Et₂O to give rise to the allylated trioxadecalin **21** as a single stereoisomer (Scheme 4).¹⁷ Asymmetric dihydroxylation¹⁸ of **21** gave a mixture of diol isomers (β : $\alpha = 3.6$:1), which were easily separated after conversion to the cyclic carbonate with triphosgene. The major product was treated with tetrabutylammonium fluoride

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⁽¹⁶⁾ Although we tried to construct directly the second ring system of the right-half segment using **14** and **15**, the desired bicyclic compounds were not formed in considerable yield.

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^{*a*} Reagents and conditions: (a) Ph₃PMeBr, BuLi, 0 °C to rt (100%); (b) liquid NH₃, Li, -78 °C; (c) DDQ (74% for two steps); (d) PdCl₂, CuCl, O₂, DMF-H₂O (83%); (e) O₃, -78 °C, then Me₂S (80%); (f) DIBALH, -78 °C; (g) Ac₂O, pyridine (96% for two steps); (h) allyltrimethylsilane, BF₃·Et₂O, 4 Å molecular sieves, 0 °C; (i) K₂OsO₄·2H₂O, dihydroquinone 9-phenanthryl ether, K₃Fe(CN)₆, K₂CO₃ (96%, $\alpha:\beta = 1:3.6$); (j) triphosgene, Et₃N (72%); (k) Bu₄NF (92%); (l) Jones reagent; (m) (PhO)₂P(O)N₃, Et₃N, 2-(trimethylsilyl)ethanol, 4 Å molecular sieves, 65 °C (70% for two steps); (n) LiOH·H₂O, MeOH (76%); (o) TBDMSCl, DMAP, Et₃N (100%); (p) CF₃SO₃Me, 2,6-di-*tert*-butyl-4-methylpyridine, 70 °C (66%).

to yield alcohol 22, which was subjected to Jones oxidation followed by a Curtius rearrangement reaction¹⁹ in the

presence of 2-(trimethylsilyl)ethanol to furnish carbamate 23. The stereochemistry of 21 and 23 was established by the NOE experiments as shown in Figure 4. Finally, to



confirm the structure of 23 by direct comparison, cyclic carbonate 23 was transformed into compound 3^{7c} in three steps.

In conclusion, a stereoselective synthesis of the right half of the mycalamides has been demonstrated by employing Lewis acid promoted intermolecular aldol reaction and oxypalladation as the key steps. This route would be adaptable to the preparation of onnamides and theopederins.

Acknowledgment. We thank Professor P. J. Kocienski for providing us with the spectral data of compound **3**.

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